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## **Ballast Water Management Practices of Foreign Flag Vessels**

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The Shipping Federation of Canada is an association of ship owners, commercial operators and agents who represent 95% of Canadian-owned oceanic traffic moving to and from ports on the east coast, the St. Lawrence and the Great Lakes. The Federation has been an active player in the implementation of ballast water management regulations for the Great Lakes, and is keenly interested in the development of a regionally coordinated approach to ballast water management that is based on internationally accepted standards and practices. In 2000, the Federation developed a "Code of Best Practices for Ballast Water Management," which is a voluntary program used by oceans ships as a means of minimizing the risk of introducing non-indigenous aquatic organisms and pathogens into the Great Lakes. The code enumerates a variety of measures that ships agree to undertake in this respect, including the cleaning of tanks, commitments with regard to areas and periods for ballast water intake, and agreement to conduct ballast water management at every practical opportunity.

Oceanic shipping is an international activity by its very nature, with ships sailing many different routes throughout the world. The masters and crews of those vessels are subject to an increasing number of standards, procedures and regulations. As a result, it is important to ensure that such rules and regulations are as simple and user friendly as possible.

Foreign-going vessels rarely have to conduct full ballast exchange along the east coast (which would take 24 to 30 hours, or some 350 nautical miles, to complete). They more frequently exchange ballast from two or three tanks (which takes 8 to 9 hours). However, coastal vessels coming from ports along the east coast or the Atlantic provinces usually find it impossible to conduct a full exchange outside the Economic Exclusive Zone (more than 200 nm from shore) due to their trading limits. Thus, such vessels have, until now, been exempted from requirements to comply with Canadian and US ballast water guidelines.

We believe that the most viable means of addressing cases in which vessels are unable to exchange ballast water at sea (due to safety considerations or coastal trade limitations) is to define a clear alternative ballast water management option. Effective development and use of such a zone or procedure would, however, require bi-national coordination.

It is extremely important that any effort to develop new measures governing ballast water management be consistent with the guidelines and regulations that are either already in place or under development at the international and national level. Legislators should also bear in mind that frequently changing limits and reporting procedures over a short period of time would lead to confusion and a reduction in compliance. A common approach from the Atlantic provinces and the States on the East Coast is absolutely essential.

Ballast water treatment systems are probably the more promising tool for effective ballast water management. The Shipping Federation supports the development of efficient treatment options and some of its members have installed trial systems on their vessels. However, at this stage, none of the systems that are available on the market have proven to be sufficiently effective to be installed on a large scale. In addition, treatment systems represent major financial investments that ship owners are not likely to undertake until an international or national standard has been developed. It is also important to remember that the installation of treatment systems on board fleets could take several years given the extensive dry-docking that might be required. Thus, given the highly competitive environment in which shipping takes place, any effort to impose a treatment requirement on vessels calling at a particular state or region would result in a diversion of cargo to other ports and threaten the economy of the region as a whole.







# Ballast Tank Biofilms as "Seed Banks": Physicochemical and Microbiological Characterization

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Lessons repeatedly learned from the past underscore the need for control of ship-surface biofouling and source ballast water. The probability of successful establishment of self-sustaining populations of exotic species is expected to increase with greater volumes of ballast water and reduced ship transit times. In addition to potential environmental and public health impacts caused by invasive species, modern aquaculture development in coastal zones is at risk of disease transfer from ballast water. More recently, the potential for ballast-tank biofilms to act as "vertical seed banks" during and after multiple ballast exchanges has become a specific concern. The overall goal of this multi-university collaboration is to assess the potential for ballast-tank biofilms to globally transport microorganisms and pathogens. Previous field experiments during transoceanic voyages demonstrated that biofilms form on multiple types of artificial surfaces deployed in ballast-water holds. When the substrata and associated biofilms were removed from the hold and submerged in sterile artificial seawater, they readily seeded secondary biofilms. In addition to bacterial populations, the primary biofilms contained a significant contingent of protista, which themselves may carry bacteria and viruses harmful to humans and other species. Continuing field experiments include deployment of controlled-surface sampling units in ships' ballast tanks, aseptic collection of biofilms from the walls of recently deballasted tanks, and laboratory experiments with a 260-gallon (1 m<sup>3</sup>) model ballast tank supplied with sand-filtered Lake Erie water. Analyses of field biofilm samples have revealed the presence of potential pathogens, including *Pseudomonas putrefaciens*, *Vibrio alginolyticus*, and *Vibrio cholerae*. The *Vibrio cholerae* detected were not representatives of human-cholera serogroups. Dinoflagellate cysts recovered from field biofilm samples later germinated to swimming cells. Contrary to our expectation of biopolymer-rich films, biofilms collected from the walls of deballasted tanks were richer in silicates and silica, as indicated by infrared spectroscopy and supported by SEM/EDXray observations of sediments. An apparent lack of substantivity (lack of adhesion to the substratum) indicated that portions of ballast biofilms can be easily stripped from tank walls under modest shear. Although not surrounded by the usual slime components of biofilms from nutrient-rich systems, as detected by infrared spectroscopy, a thriving microbial community was revealed by light microscopic and immunofluorescence analyses of ballast biofilms. When replicates of these test plates were subjected to controlled shear conditions (stagnation point flow cell apparatus), much of the micro-particulate and microbial load was removed. Subsequent immunofluorescence analysis of the shear-challenged samples, however, showed that populations of target organisms were still present on the surface. These, more strongly retained, organisms can survive in the biofilm to produce new generations of cells for later release into the ballast water. In the model ballast tank containing Lake Erie water seeded with ballast-tank sediment, biofilms formed readily on all test surfaces. Preliminary light microscopic analyses of these biofilms revealed bacteria and protista. Species identifications are underway.

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# Evaluations of Deoxygenation as a Ballast Water Treatment to Prevent Aquatic Invasions and Ship Corrosion

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One of the most important mechanisms for the introduction of aquatic nuisance species is transport in ship ballast waters. Although several ballast tank treatments to prevent transport of aquatic organisms appear promising, all existing approaches will result in significant costs to the shipping industry. The implementation of ballast water treatment measures would be hastened by providing the shipping industry with economic incentives for doing so. Our initial work suggests that deoxygenation may be such a treatment with benefit for ship owners by reducing corrosion, while simultaneously limiting the number of aquatic organisms surviving transport in ballast tanks.

Purging of oxygen from ballast tanks with nitrogen gas was recently proposed by Sumitomo Heavy Industries to be a cost-effective technique for reducing corrosion and therefore extending ship life. An initial shipboard study found a 10-fold reduction in corrosion rates under hypoxic ballast tank conditions. In previous studies, we also tested both the oxygen tolerance of larvae from known invasive invertebrate species and conducted literature reviews on the ability of aquatic organisms to survive low oxygen environments. These investigations suggest that few organisms will be able to withstand extended periods of exposure to hypoxic ballast water.

While initial results are promising, several questions must be addressed before definitive conclusions can be drawn regarding the feasibility of deoxygenation as a ballast water treatment. In particular, issues of cost effectiveness, responses of natural planktonic communities, microbiologically influenced corrosion, and scaling up to shipboard applications require additional investigations.

We will discuss the potential of deoxygenation to prevent both invasions and corrosion by briefly reviewing our past work and reporting on our current investigations to: 1) optimize the oxygen stripping process (with particular emphasis on the ballasting flow rates and tank volumes of large vessels); 2) determine mortality levels of natural Chesapeake Bay planktonic communities (zooplankton, phytoplankton, and microbes) under hypoxia in mesocosm experiments; and 3) quantify corrosion rates and establish the mechanisms under deoxygenated conditions (with particular emphasis on microbiologically influenced corrosion). These steps will ultimately lead to a full-scale study to determine the efficacy of deoxygenation at removing ballast water organisms onboard active vessels.

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## Large-Scale Testing of Hydrocyclone, Screen and UV Radiation as Ballast Water Treatment Technologies

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Large-scale dockside experiments were undertaken to evaluate the treatment efficiency of commercially available unit processes for preventing the transfer of unwanted species via ships' ballast water. The project was undertaken with funds from the US Coast Guard Research and Development Center, and the treatment system was built at the University of Miami's Rosenstiel School of Marine and Atmospheric Science campus. The test water from Biscayne Bay, Florida, was pumped through the system at approximately  $5.7 \text{ m}^3 \text{ min}^{-1}$  (1500 gpm or gallons per minute). The unit processes included a hydrocyclone (centrifugal force of approximately  $13 \times g$  at  $5.7 \text{ m}^3 \text{ min}^{-1}$ ), a self-cleaning  $50 \mu\text{m}$  (micron) screen, and a UV (ultraviolet) treatment unit (100% transmittance =  $60\,000 \mu\text{W s cm}^{-2}$ ). In addition to these unit processes, a mixing system, including a high pressure injection pump, was fabricated to add suspended solids (clay slurry) to the flowing Biscayne Bay water to enhance turbidity. The effect of increased suspended solids and enhanced water color on the unit processes was the main focus of this research.

A broad spectrum of biological and biochemical indices were monitored to evaluate treatment efficiency. During each experimental run, 760 L (200 gal) samples were collected and passed through  $35 \mu\text{m}$  plankton nets for zooplankton collection. Water samples were also collected for phytoplankton, microbiological, ATP and protein analyses. After the initial samples were collected, a second set of samples was held for 18 hours (turbidity experiments) or 6 to 7 days (color experiments) to determine the effects of storage on the effectiveness of treatment processes. Statistical analyses showed that hydrocyclonic separation was not effective for treatment. The 50 mm screen, however, was effective at removing over 90% of the zooplankton species monitored, and a small percentage of the microphytoplankton in the natural water samples. Initially, UV treatment was able to reduce the count of viable microorganisms to an undetectable level; however, bacterial regrowth was observed in samples held for 18 hours. There was no regrowth in samples held for 6 to 7 days.

The main variable explored in this research was the effect of increased suspended solids (turbidity) or enhanced color on the unit processes. While the dose delivered by the UV system decreased due to the increased suspended solids loading, its reduced value (approximately  $25\,000 \mu\text{W s cm}^{-2}$  minimum dosage) was still sufficient to inactivate microorganisms, even with the increased turbidity. In subsequent experiments, a UV dose of  $10\,000 \mu\text{W s cm}^{-2}$  was found to be insufficient to inactivate microorganisms.

In general, it was observed that only the 50 mm screen contributed appreciably to removal of organisms, especially zooplankton in the test water of this facility. If organisms smaller than 50 mm are to be removed or killed however, then either a more efficient filtration process or a biocide will need to be added to achieve these treatment goals. It appears that UV treatment will not be effective at facilitating any meaningful treatment, especially against potential invading organisms such as large zooplankton or larvae of invertebrates.





# Kill Harmful Microorganisms in Ships' Ballast Water Using Hydroxyl Radical

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A physics method is studied that the electrons are accelerated and then the gas molecules are aroused using a strong dielectric barrier discharge. With this method, the strong electric field ( $E_d \geq 400 \text{Td}$ ,  $1 \text{Td} = 10^{-17} \text{Vcm}^2$ ) is formed with the thinner  $\alpha\text{-Al}_2\text{O}_3$  dielectric layer in the narrow discharge gap at a high pressure ( $P \geq 0.1 \text{Mpa}$  or  $n = 2.6 \times 10^{-19} / \text{cm}^3$ ). The electrons achieve the average energy of above 12eV. The ionization potential of  $\text{O}_2$  and  $\text{H}_2\text{O}$  molecules is 12.5 eV and 12.6eV respectively. The ionization numbers of  $\text{O}_2$  in strong ionization discharge is 12 769 times higher than that of  $\text{O}_2$  in other gas ionization discharge at normal pressure. As a result,  $\text{O}_2$  in air and  $\text{H}_2\text{O}$  in seawater are ionized and dissociated into a number of activate particles such as OH,  $\text{O}_2^+$ , O ( $^1\text{D}$ ),  $\text{HO}_2$  radical and so on. High dissolved hydroxyl radical ( $\text{OH}^*$ ) concentration of more than 10mg/L is achieved by the activate particles dissolved into seawater (a part of ballast water) with high mass transfer efficiency. 1.0 t dissolved hydroxyl radical is enough to treat about 10.0 t ballast water. The hydroxyl radical is a "green" ideal medicament without any toxin and residues, which is possible to be decomposed into  $\text{H}_2\text{O}$  and  $\text{O}_2$  after 20 minutes.

The principles of organisms killed by hydroxyl radical are as follows:

- 1) Oxidization and decomposition of amino acids of introduced organism. The activated groups such as -OH, - $\text{NH}_2$  guanidino etc in amino acids of protein play an important action on maintaining proteinic configuration and catalytic activity of enzyme. When the hydroxyl radicals react with the activated groups, the chemical damage of protein further the death of introduced organisms occur.
- 2) The hydroxyl radical reacts with Deoxyribonucleic Acid (DNA) to produce DNA adducts resulting in a chemical damage without any restoration.
- 3) Hydroxyl radicals acts on the lateral chain of phosphatide polyunsaturated fatty acid of cell membrane to lead to the fast degradation of polyunsaturated fatty acid further to the damage of cellularity and then to cells death. In a world, the hydroxyl radical has a broad-spectrum deadly characteristic that is possible to kill microorganisms meanwhile bleaching and deodorization. The hydroxyl radicals act on the introduced microorganisms belonging to a dissociative radical reaction with a fast reaction rate.

When the dissolved hydroxyl ratio concentration is 0.15mg/L, the kill efficiency of bacteria sum is 99%. When the ratio concentration is 0.25mg/L, the kill efficiency of bacteria sum reaches 100%. After 5 minters of reaction, the hydroxyl ratio concentration to kill *Ba. Navicula*, *Ch. Dunaliella*, *P. Subcordiformis* is in range of 0.6~0.8 mg/L respectively, and the kill efficiency is above 99.0%. When the ratio concentration is 0.9 mg/L, 1.16mg/L, 1.12mg/L, the kill efficiency reaches 100%. The algae chlorophyll-a change after 24 hours of hydroxyl reaction is shown in Fig.4. The hydroxyl ratio concentration of suppressing chlorophyll-a increase is in the range of 0.21~0.34mg/L, the ratio concentration of all chlorophyll-a decomposed is in the range of 0.55~0.68mg/L. Therefore the dissolved hydroxyl value to kill microorganisms is 0.8mg/L.

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# Technology Policy Implications for Ballast Water Introductions of Non-indigenous Species

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The problem of non-indigenous species introductions only escalates, as the world becomes a global economy. Increased trade between different locations provides new vectors of introduction for many non-indigenous species. The most common method of introduction has been identified as ballast water from ships. The scientific community is currently trying to evaluate the economic, ecological, and human impacts of these introductions, but it is increasingly important that policy analysts become involved in the research. There is currently not a model that can determine the likelihood that an introduced species will become invasive. It therefore is necessary to evaluate the potential of reducing the risk factors that lead to introductions. Policy makers at the national and international level are currently struggling to formulate and implement policies that will effectively reduce the risk of introductions to coastal waters. The proposed policies advocate a global standard that may not adequately protect the environment and may impose unnecessary restrictions or financial burdens on the shipping industry.

The risk reduction model developed through this research evaluates both the likelihood of invasions from oceanic voyages as well as coastwise shipping introductions. The model characterizes three elements: 1) the port of origin environment (e.g., salinity, temperature); 2) the voyage factors (e.g., treatment method, length of voyage, volume of ballast, and ballasting events); and 3) the port of destination environment to determine the risk reduction potential of technology policy combinations. This work evaluates the risk of species being introduced and then extrapolates to produce an estimate of invasion risk.

The model will be applied to a variety of ports in the United States, representative of geographic location, vessel type, cargo volume, and ecological conditions. The model is intended to evaluate the available policy and technology solutions to determine the most effective combination to reduce the risk of introduced species to individual port environments. The model will illustrate the dominant factors of introduction and the areas of greatest concern. Information from the model will enable policy makers to implement policies that address specific factors and trade routes, instead of the broad issue of invasions.

The risk reduction model can be applied to any port to determine which policy and technology combination will most effectively combat the risk of introduction of invasive species. The model developed by this research will inform the policy making process by evaluating the factors that are controllable by regulations, technology and policy requirements. Previous research and models in this area have address the biological conditions necessary for invasion, these cannot be changed by regulations, therefore it is necessary to develop a model that involves the ports and ships, the two areas that can be regulated. This presentation will discuss the development of the model and explain the factors evaluated by the model, and preliminary results from the Port of Houston, Texas.

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# A Pilot-Scale System for the Treatment of Ships' Ballast Water Using Hydroxyl Medicament

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The high dissolved hydroxyl concentration is formed using the strong dielectric barrier discharge and its kill principles are introduced in paper, the title "Kill Harmful Microorganisms in ship's Ballast Water Using Hydroxyl Radical". In this paper, a pilot-scale system for the treatment of ship's ballast water and its experimental results are discussed.

High-concentration hydroxyl radical is produced in Hydroxyl generator 3, and then is injected into dissolved unit 8 to form high dissolved hydroxyl medicament. The ratio concentration of OH\* medicament reaches the threshold value of 0.65mg/L in the main pipeline of the ballast water. The introduced microorganisms are killed along a 6 m long main pipe, 10 m long is enough. In this system, the flow rate of ballast water is 20 t/h, the flow velocity is 1.6 m/s, the sample point is along 6 m long of main pipe.

The ship's ballast water in this system is taken from the Dalian port, Liaoning Province, P.R. China. The Phytoplanktons, Zooplanktons and Germs are as follows. Phytoplankton: Chlorella Pyrenoidosa, Chaetocers, Peridinium. Zooplankton: Euplotes. Germ: Pseudomonas, Flarobacterus, Vibrio, Acinetobacter, Escherichia, Alcaligenes, Staphylococcus.

Because of low content of the monoplast algae and germs in ship's ballast water, 2216E liquid nutrient medium is injected into a glass tank of 1.5m<sup>3</sup> for the enrichment culture of the algae and germs which are enriched into 2<sup>o</sup>;10<sup>12</sup>/m<sup>3</sup> and 4<sup>o</sup>;10<sup>6</sup>/m<sup>3</sup> respectively. When the ratio hydroxyl

concentration is 0.65 mg/L, the germs, monoplast algae and Zooplanktons are not tested in the water sample of 150 mL. The introduced microorganisms can be killed within 4.1 s also along the main pipe of the ballast water.

The conclusions that hydroxyl radical kills the introduced microorganisms is as follows.

1) Lipide peroxidation degree of monoplast algae increases 2 times in the experiment resulting in death.

2) Hydroxyl radical has a strong destruction effect on monosaccharide, amylose, protein, DNA and RNA in the plankton cell.

3) Hydroxyl radical has a obvious destruction to the activity of CAT, POD and SOD in antioxidase system of Plankton. Therefore, the first is the loss of enzyme activity. And then, the enzyme as a protein is oxidized and decomposed by hydroxyl radical. Especially, antioxidase system is destroyed, which effect vital activity of the cells so that the cell die.

The content of phosphate and nitrate has a little raise of 11.5% and 17.4% respectively. The content of nitrite and ammonium salt has a obvious decrease from 66.6 ug/L to 1.1 ug/L and 79.8 ug/L to 0.4ug/L respectively, which are oxidized and decomposed by hydroxyl radical. The products have Nitrate to lead to a little increase of Nitrate. Nitrite and ammonium salt have a harmful effect on aquatic, especially Nitrite can oxidized the low-iron hemoglobin into the high-iron hemoglobin resulting in the loss of function to carry oxygen. Therefore the water quality is obviously improved after the treatment of hydroxyl radical.

The decrease of 50% turbidity, no COD tested, 2 times raise of TOC indicated that hydroxyl radicals oxidize and decompose the organics in water into COD and OC is difficult to be tested. In addition, The oxygenolysis of hydroxyl radicals to the cell remains also contributes to the increase of TOC content.

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